Common Sources of Anions

NO_3^- (Nitrate)	gunpowder, matches, fertilizer
$C_2H_3O_2^-$ (Acetate)	vinegar, buffers, photography, tanning
Cl ⁻ (Chloride)	table salt, seawater, photography, food preservative
Br ⁻ (Bromide)	seawater, sedatives, photography, some fire extinguishers
I ⁻ (Iodide)	iodized table salt, tincture of iodine (I_2/NaI mix)
SCN ⁻ (Thiocyanate)	colorimetric iron test reagents, photography
S ²⁻ (Sulfide)	ores, eg. pyrite, FeS (fool's gold), PbS (galena), egg yolks
SO_4^{2-} (Sulfate)	MgSO4 (Epsom Salts), fertilizer, wallboard, plaster, cement
CO ₃ ²⁻ (Carbonate)	CaCO ₃ (limestone), Na ₂ CO ₃ (washing soda), antacids, pigments
CrO ₄ ²⁻ (Chromate)	anti-rust paints, artist pigments, chrome plating solutions
PO ₄ ³⁻ (Phosphate)	fertilizer, water softening, baking powder, buffers, bones
AsO ₄ ³⁻ (Arsenate)	rat poision, insecticide, tanning, some Pb ores
AsO ₃ ³⁻ (Arsenite)	rat poision, insecticide, manufacture of mirrors, some Pb ores
F⁻ (Fluoride)	insecticide, fluxes, enameling, fluoridation of water, toothpaste
CN⁻ (Cyanide)	gold & silver ore extraction, electroplating, poisons
NO_2^- (Nitrite)	dye manufacture, meat preservative, corrosion inhibitors
HCO_3^{-} (Bicarbonate)	NaHCO ₃ (Baking Soda), groundwater
SO_3^{2-} (Sulfite)	wine preservative, photographic developers, bleaching
BO ₃ ³⁻ (Borate)	eyewash, tanning, wood preservative, fire retardant, flux
PO_3^{3-} (Phosphite)	fertilizer

F⁻, Cl⁻, Br⁻, and I⁻ ions are called halides or halide ions. F_2 , Cl_2 , Br_2 , I_2 as elements, are called halogens.

SCN⁻, thiocyanate ion, is called a pseudohalide, because its chemistry mimics the halides.