

FS/2019
ORGANIC CHEMISTRY I LAB
CHEM 226 (2219)

Instructor:

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<http://www.mst.edu/~tbone>
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OBJECTIVES:

Course objectives are to develop facility with performing laboratory techniques involving the handling of organic chemicals safely and the keeping of proper records of experiments conducted in the laboratory. Crystallization, distillation, extraction and chromatography are emphasized as separation and purification techniques. Melting points, boiling points, and refractive indices are used routinely as measures of purity. Safe handling of chemicals and proper disposal of waste products are priority goals in this course. To help in attaining such goals, the first week of the semester is devoted to SAFETY in the chemistry laboratory. For the remainder of the semester, most of the experiments are conducted on a micro scale.

BOOKS:

1. 100 pg. set Organic Chemistry Laboratory Notebook
2. "Organic Chemistry Laboratory Laboratory Techniques, 2nd ed". Available as free pdf download at <https://open.umn.edu/opentextbooks/BookDetail.aspx?bookId=369>
3. OPTIONAL "Microscale Techniques for the Organic Laboratory, 2nd ed", (MTOL), Mayo, Pike, Butcher and Trumper, John Wiley & Sons, Inc., 2001 (on reserve at library circulation desk-currently out of print but available on Amazon)
4. OPTIONAL "Right to Know pocket guide for School & University Employees", Genium Publishing Corp., 1990. (optional)

(#1 can be purchased from the book store or Amazon)

LAB SYLLABUS:

Originally prepared by Prof. S. B. Hanna, modified by D. E. Hoiness, T.Bone.

GENERAL GUIDELINES

SAFETY

Goggles must be worn at all times in the lab.

Nitrile gloves (blue or purple) and lab aprons are recommended but not required. You may want to purchase nitrile gloves (blue or purple) from the bookstore or Walmart. Do NOT use vinyl or latex gloves as these provide inadequate protection.

Shorts or short skirts and open toed shoes are not permitted.

MSDS info is available online via the class web page or at <http://ehs.mst.edu/msds/MSDS.html>.

NOTEBOOK

1. A 100 pg. Computation Notebook, with white and yellow pages, available at the bookstore, should be used. These are designed so that a carbon copy is made on the yellow pages which can then be torn out and turned in to the grader on completion of the experiment. Be sure to press firmly so that the yellow copy is readable. If you have bad handwriting, you may want to print.
2. ALL entries in your notebook should be recorded in permanent blue or black ink. NO pencil.
3. **Table of Contents:** Two pages should be retained at the front of the notebook for the table of contents, to be updated weekly.
4. Print at the top of every page. Your name, Course and Section number, Date, **Title** of the Experiment
5. Prelab: This portion must be completed before you come to lab lecture. Prelabs are due when you walk in to lecture and will be considered late upon start of the lab lecture.
Objective: Brief summary of the objective of the experiment.
Chemical Equations: (if applicable)
Physical Properties: For all chemicals used in the experiment, list as a table,

Compound name	Structure	CAS#	MWt.	BP or MP lit. °C	State (s, l) color	RI (liquids only)	Hazards

Property data may be found in the Merck index, CRC, Aldrich catalogs or online. These are available at the library circulation desk or my office, rm 120B.

Reference(s) should follow the table as to the source of the property data.

Prelab Questions: answer any assigned questions in Canvas online.

6. **Procedure:** This section must be written as you do the lab. It should include experimental data and an apparatus sketch. It should be complete enough to allow someone to repeat the experiment and should note any procedural modifications used. (Use 3rd person past tense)
7. **Observations:** Report what you witnessed concerning the reaction, such as color changes, ppt, gas, etc.
8. **Results:** Results should be reported in the form of a table. The yield and physical properties may be listed as two separate tables if more space is required. Show theoretical yield, % yield and % error calculations below the table.

Product name or unknown #	Yield(gm) exp.	Yield (gm) theor.	% Yield	MP, BP, RI exp.	MP, BP, RI lit. (ref)	% Error MP, BP, RI

9. **References:** Cite source(s) used for product physical property data in table.
10. Each section (5-9) must be clearly designated.
11. Each new experiment should begin on a new page.
12. Void all unused space. Sign and date each page.
13. **You must have your lab book initialed by the TA before leaving the lab.**
14. To make corrections: simply draw a line through the old data or conclusions, add the new information and initial it. Do not erase or use white out.
15. Upon completion of each experiment, the yellow pages of your notebook are to be turned in to the grader before the end of the next lab session. This will allow time to complete MP or yield measurements if not finished the previous period. Late reports will receive only partial credit. (-5 pts / wk late, -25 pts max)

Notebooks will NOT be graded if: pencil is used, white-out is used, or a TA did not sign your notebook at the end of the day.

CELL PHONES:

Cell phones are to be turned off and put away during lecture. Any use of cell phones during lecture will result in confiscation of the phone for the remainder of the lecture.

MAKE-UP OF LABS:

Make up labs must be completed within one week of the original scheduled date, as the chemicals for each lab are put away after each experiment. Exceptions may be made for a written medical excuse or with the instructor's permission. If you know in advance that you will have to miss a lab, please notify your TA and instructor. With advance warning, we may be able to fit someone into another lab section that week. Missed labs remaining unexplained after their due date will receive no credit and cannot be made up.

The lab schedule is online at: <http://web.mst.edu/~tbone/Subjects/TBone/scheduleFS2019.html>

GRADES

Weekly quizzes will be given at the end of each lab lecture covering general knowledge of the experiment you will be doing that week. Quiz topics may include compound names or structures, reactions, equipment, or techniques used.

Quiz

15 pts each

Lab Notebook

10 pts for CER prelab questions (due at beginning of lecture)

20 pts Prelab (due at beginning of lecture, see general guidelines for details)

40 pts Procedure and Observations (to be recorded while in lab)

25 pts Results (Table(s))

Total pts per experiment: ~110 pts

Penalties (-5 each, -25 pt maximum)

Procedure not in 3rd person past tense

Improper method of calculation

Missing or incorrect data or results

Failure to void unused space

No references for properties

Results not in table form

Failure to sign and date each page

Turned in late (-5 pt/wk)

Weekly Experiments are due before the end of lab the week following the completion of the experiment.

Exams, (Midterm and Final) are both required and will have -5% / day deducted from the curved score if taken after the scheduled section time.

Course

9 experiments	765 pts	59%
10 Canvas prelabs	100 pts	8%
9 highest quizzes	135 pts	10%
Mid term exam	100 pts	8%
Final exam	<u>200 pts</u>	<u>15%</u>
Total	1300 pts	100%

Course grades will be based on the following % scale. (Some curving of raw scores may be applied.)

90-100	A
80-89	B
70-79	C
60-69	D
<60	F

GENERAL LAB SUPPLIES

Required Items: Students are required to supply their own goggles and paper towels.

Safety Items Dial 911 for emergencies.

You are responsible for knowing the location of the following safety items in the lab: Mark these on your lab map, included with this handout.

Safety shower, eyewashes(3), fire extinguisher

Balances are to be kept clean. A pan and broom for spill cleanup are by the supply cart. Dispose of used weighing paper, etc. in the used solids waste bucket in the waste hood.

Chemicals & Samples for each lab are located under the hoods or by the balances.

Supply Cart –in lab

1.5 ml latex bulbs (reuse these)
Pasteur pipets, (9" glass, dispose in glass waste, after rinsing)
Sample Vials, (Snap Cap , 2 ml)
Corks/Stoppers
Labels (S, M, L)
Q tips
Filter paper, Phase separation paper
pH paper, litmus paper
Aluminum foil, Parafilm
Cotton batting (insulation)
Scissors, stapler
Stopcock grease
Magnetic stir bar retriever (micro stir bars are \$11.00 ea!)

Waste Hood PLEASE REPLACE THE LID ON ALL WASTE CONTAINERS AFTER USE

Non Halogenated Solvent Waste
Halogenated Solvent Waste (compounds containing F, Cl, Br, I eg. CH_2Cl_2)
Mineral Acids Waste-(aqueous strong acids and bases)
Glass Waste (box for Pasteur pipets, snap cap vials, broken glass, MP cover glasses)
Used Solids Waste-(white bucket, for contaminated filter paper, etc. -NO glass items)
Mercury Waste (broken thermometers)
Sharps Waste (syringe needles)
Wash bottles of acetone, water, (for cleaning glassware).

Hoods Large lab hoods will sound an alarm if the sash is raised too high. There is a reset button on the upper right front that will temporarily disable the alarm. The sash should be returned to a 4"-6" opening when finished.

Mini hoods at each lab bench location should be used for all experiments. There is a flow shutoff valve on each that must be opened before use and closed after use.

Spills & Breakage Cleanup Broken glass should be disposed of in the glass waste box, which is located in the waste hood. There is a pan and broom, kitty litter, baking soda etc. available in lab by the supply cart. Please see the TA for assistance.

Students are responsible for all breakage, loss or other damage to equipment. Damaged items must be repaired, replaced or paid for before check out is complete.

Breakage Payment-Contact the TA to fill out a green slip and obtain a replacement item. Payment is via charge to the student account.
Be sure to obtain a cash receipt from the TA when paying.

Failure to check out will result in a \$25.00 fine. You must check out even if you drop the course.

GROUP STATION CABINET CONTENTS

Top Shelf

- 4 Filter Flask, 250 ml
- 2 Filter Paper, 3.0 cm dia (for Hirsch Funnel)
- 4 Hirsch Funnel, 3.0 cm dia
- 4 Mortar and Pestle
- 4 Test Tube Rack with 6 Test Tubes
- 4 Vacuum Trap

White Plastic Tray (bottom shelf)

- 4 Beaker Tongs
- 4 Crucible Tongs
- 4 Cu Wire, heavy gauge
- 4 Ring Clamp, Small
- 8 Test Tube Clamp/Holder
- 4-Tweezers(forceps)

Bottom Shelf

- 1 Aluminum Heating Block
- 4 Steam Bath, Cu
- 4 Vacuum Hoses (1/2" O.D.-heavy wall)
- 8 Water/Gas Hoses (3/8" O.D.-thin wall)

The above items are shared by all sections and must be returned to the common drawer after use.

DESK CONTENTS

Microscale Kit Components

- 1 Air condenser
- 1 Jacketed condenser
- 1 Claisen adapter
- 1 Hickman still
- 1 Drying tube
- 1 5.0 ml conical vial
- 2 3.0 ml conical vial
- 1 1.0 ml conical vial
- 1 0.1 ml conical vial
- 1 Teflon spin vane-large
- 1 Teflon spin vane-small
- 2 2 ml GC sample vial/cap
- 1 Plastic 1 ml syringe
- 1 Micro filter paper in glass vial, 0.5 cm dia

Extra Components

- 2 Watch glass, 3" dia
- 1 Vacuum filtering flask, 25 ml
- 1 1 cm Hirsch funnel with adapter
- 1 Casserole
- 2 Erlenmeyer flask, 50 ml, 125 ml
- 3 Beakers, 50 ml, 100 ml, 150 ml (1 ea)
- 1 Short stem glass funnel
- 1 Thermometer
- 1 Glass stirring rod with policeman
- 1 Plastic Ruler, 6"
- 1 Glass syringe, 2 ml
- 1 Graduated cylinder, 10 ml

Opening combination locks: Turn the dial 3 turns clockwise and stop on the first number of the combination. Next, turn counterclockwise, passing the middle number once and stop on the middle number of the combination the second time it comes up. Finally, turn clockwise and stop on the last number of the combination. The locks will not open if you miss any number by more than one digit.

Be sure to record your desk number and combination on something that you will be bringing to lab after check in.

Disability Support Services

If you have a documented disability and anticipate needing accommodations in this course, you are strongly encouraged to meet with me early in the semester. You will need to request that the Disability Services staff send a letter to me verifying your disability and specifying the accommodation you will need before I can arrange your accommodation.

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TITLE IX Info.

Missouri University of Science and Technology is committed to the safety and well-being of all members of its community. US Federal Law Title IX states that no member of the university community shall, on the basis of sex, be excluded from participation in, or be denied benefits of, or be subjected to discrimination under any education program or activity. Furthermore, in accordance with Title IX guidelines from the US Office of Civil Rights, Missouri S&T requires that all faculty and staff members report, to the Missouri S&T Title IX Coordinator, any notice of sexual harassment, abuse, and/or violence (including personal relational abuse, relational/domestic violence, and stalking) disclosed through communication including but not limited to direct conversation, email, social media, classroom papers and homework exercises.

Missouri S&T's Title IX Coordinator is Vice Chancellor Shenethia Manuel. Contact her directly (manuel@mst.edu; (573) 341-4920; 113 Centennial Hall) to report Title IX violations. To learn more about Title IX resources and reporting options (confidential and non-confidential) available to Missouri S&T students, staff, and faculty, please visit <http://titleix.mst.edu>.

FILES

There are numerous files of prelab & postlab questions circulating that have incorrect answers, although due to previous careless grading, were marked as correct. Incorrect answers will not be given credit, so you should not rely on files as a source of easy answers.

CELL PHONES

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