Lecture 12: Potential energy diagrams

- Problem with Work done by "other" forces
- Relationship between force and potential energy
- Potential energy diagrams

Example with other force



A block of mass M is at rest on an incline that makes an angle θ with the horizontal. It slides down a distance L and then flies of the edge a height H above the ground. Throughout its motion, a constant vertical blowing force of magnitude B is acting on the block.

Derive an expression for the speed with which the block hits the ground.

Relationship between force and potential energy

$$U(\vec{r}_B) - U(\vec{r}_A) = -W_{A \to B} = -\int_{\vec{r}_A}^{r_B} \vec{F} \cdot d\vec{r}$$

In one dimension: $\vec{F} = F_x(x)\hat{i}$

 $\Delta U = -\int F_x dx$

$$F_x = -\frac{dU(x)}{dx}$$

$$U_{grav} = mgy$$
 (y-axis up) $F_{grav,y} = -\frac{d}{dy}(mgy) = -mg$

In three dimensions: U = U(x, y, z)

$$F_x = -\frac{\partial U(x, y, z)}{\partial x}$$
 $F_y = -\frac{\partial U(x, y, z)}{\partial y}$ $F_z = -\frac{\partial U(x, y, z)}{\partial z}$

Partial derivative $\frac{\partial}{\partial x}$ means: treat y and z like constants and only x like a variable

Example:
$$U(x, y, z) = xy^2 z$$

 $\frac{\partial u}{\partial x} = y^2 z$, $\frac{\partial u}{\partial y} = 2xyz$, $\frac{\partial u}{\partial z} = xy^2$

Motion in a potential energy well

Consider motion in one dimension under the influence of a single conservative force with potential energy U(x).

 $W_{other} = 0$, $E_f = E_i$



Kinetic and potential energy



$$E = K(x) + U(x)$$
$$\implies K(x) = E - U(x)$$

Small $U \Longrightarrow$ large KLarge $U \Longrightarrow$ small K

K is maximum where *U* is minimum

At turning points: U = E, K = 0

Force and potential energy



Different total mechanical energy



Motion possible between x_1 and x_2 or between x_3 and x_4 Not possible between x_2 and x_3 because U(x) > E(*K* can not be negative)

Equilibrium



Equilibrium:
$$F_x = 0$$

$$F_x = -\frac{dU(x)}{dx} = 0$$

U(x) has local minimum or maximum

Minimum = stable



Maximum = unstable



Example: Diatomic Molecule

Diatomic molecule: two atoms separated a distance r What should we expect about force between atoms?

Too close: repulsive force

The shorter the distance, the greater the force

Too far: attractive force

Force decreases with distance, goes to zero for $r \rightarrow \infty$

Distance just right: equilibrium

Example: Diatomic Molecule

Diatomic molecule: two atoms separated a distance r

