

Physics 4311: Thermal Physics - Homework 12

due date: Tuesday, April 25, 2023, please upload your solution as a pdf on Canvas

Problem 3: Ideal gas in the gravitational field (20 points)

Consider a classical ideal gas of N particles at temperature T in a vessel of cross section A and height H . The particles of the gas are under the influence of a gravitational potential $E_{pot} = mgz$ where m is the mass of a particle, g is the free fall acceleration and z is the vertical coordinate. (Assume H to be large, $mgH \gg k_B T$)

- Calculate the partition function and the free energy of the gas.
- Determine the internal energy U and the specific heat C_V . Compare with the equipartition theorem.
- Calculate how the particle density $n(z)$ changes with z . (Hint: the particle density $n(z)$ is a reduced probability density of the phase space density $p(\vec{r}, \vec{p})$.)
- What is the approximate ratio between the air density at sea level and at an altitude of 8000m? Use the following approximate values: $g \approx 10\text{ms}^{-2}$, $m \approx 4 * 10^{-26}\text{kg}$ (mass of a nitrogen molecule), $k_B \approx 4/3 * 10^{-23}\text{J/K}$, $T = 300\text{K}$.

Problem 2: Ultra-relativistic classical ideal gas (20 points)

Consider a gas of N non-interacting, indistinguishable, classical particles at temperature T in a cubic box of linear size L . The energy-momentum relation is ultra-relativistic, $E = c|\vec{p}|$, where c is the speed of light.

- Calculate the partition function and the free energy of the gas.
- Calculate the pressure as function of N , T , and V .
- Find the internal energy U and the specific heat C_V at constant volume.
- Also determine the specific heat at constant pressure, and compare the ratio of to that of the nonrelativistic case.